🖰 CHEMISTRY 🛛 🔛 MATERIALS 📶 ANALYTICAL

DualSeal Airtight-Double-Cap Bottled Reagents



Volatile Solutions Metallic Salt Solutions Dehydrated Solvents Organometallic Reagents Other Reagents

DualSeal Airtight-Double-Cap Bottled Reagents

TCI introduced our newly developed double cap system "DualSeal" for moisture/oxygen-sensitive products, allowing you to keep them in good condition until the last drop. We will continuously increase our usage of DualSeal across our product portfolio.



Features of DualSeal

- Highly airtight double cap structure.
- Air-sensitive reagents can be safely dispensed without exposure to air.
- Even after piercing the septum cap, the PTFE sheet on the blue outer cap provides sealing protection.
- · Easy to dispose of caps and bottles separately.



Outer cap (Blue)

DualSeal specification

DualSeal consists of two parts: the blue outer cap and the white septum cap, both of which can be screwed on and off.

No additional sealing is required after piercing the septum cap with a needle. Just screw the outer cap back in place.

The outer cap has a convex structure on the inside. By filling the space where air and moisture stay, the material is protected from moisture and oxygen even after piercing the septum cap with a needle.

Caution

Do not dispose of the inlaid PTFE sheet! Keep it as it is during use! This PTFE sheet acts as a second layer air seal.





Septum cap (white)

The white septum cap has a wide septum surface for ease of use, and features two layers of rubber and a highly chemically resistant PTFE seal. The septum cap body is made of polypropylene and contains a screw thread allowing for easy removal from the bottle. The septum cap is closed with high torque to ensure an airtight seal. Open the septum cap only when all of the liquid has been used up and you want to prepare for the disposal of the bottle.

Caution

Do not place anything on the septum cap! This will significantly reduce seal quality. The cap and bottle can be separated for disposal. It is easy to dispose of caps and bottles separately. Highly reactive reagents may be residing inside the cap. Please take the necessary precautions to avoid accidents due to exposure to oxygen or moisture. Dispose of the bottle and the caps separately after ensuring that no chemical residue is left behind.

How to use DualSeal

In case using a needle (1): Solvents except for Halogenated hydrocarbon solvents

- 1. Clamp and secure the reagent bottle before opening.
- 2. Carefully unscrew the blue outer cap only. Place the cap near the bottle while in use.
- 3. To prevent air from entering the container, insert a needle with a balloon filled with an inert gas such as argon or nitrogen. Then insert a syringe needle through the septum surface of the septum cap.
- 4. Fill the syringe with the required amount of liquid.
- 5. Remove the syringe, inject the liquid into your reaction vessel and safely dispose of the needle.
- 6. Take the outer cap and screw it tightly back in place.

*Repeated use can lead to increasing the number of holes or increasing the size of existing holes and will over time lead to an increase in air leakage. To prevent deterioration of the septum, reduce the number of injections as much as possible, or purchase a smaller sized bottle.



In case using a needle (2): Halogenated hydrocarbon solvents such as dichloromethane

- 1. Clamp and secure the reagent bottle before opening.
- 2. Carefully unscrew the blue outer cap only. Place the cap near the bottle while in use.
- Attach a needle to the PTFE tube for liquid delivery connected to the reaction vessel filled with the inert gas. Puncture the septum with the needle so that the needle tip reaches the bottom of the reagent bottle.
- 4. A syringe filled with an inert gas or a needle connected to an inert gas pump punctures the septum, and the inert gas is sent into the void of the reagent bottle to send the liquid.
- 5. Remove the needle of the liquid feeding tube first, and then remove the needle that was feeding the inert gas.
- 6. Take the outer cap and screw it tightly back in place. However, it is recommended that the minimum number of removals be made, as the holes will be degraded by the vapor of the halogenated hydrocarbon solvent.

How to remove the septum cap

Open the septum cap only when all of the liquid has been used up and you want to prepare for the disposal of the bottle.

- 1. Clamp and secure the bottle before opening.
- 2. Open the septum cap by unscrewing. (The septum cap is tightly closed with high torque to ensure airtightness. Using tools such as water pump pliers is recommended.)

*Do not remove the outer cap when you remove the septum cap. Take extra care to avoid any spillage of inner liquid.

Notes on bottle disposal

The cap and bottle can be separated for disposal.

- Remove DualSeal by referring to "How to remove the septum cap".
- Highly reactive reagents may be residing inside the cap.
 Please take the necessary precautions to avoid accidents due to exposure to oxygen or moisture.
- Dispose of the bottle and the caps separately after ensuring that no chemical residue is left behind.

DualSeal Sealability Test: Moisture Analysis

In order to evaluate the sealability of DualSeal, we periodically measure and monitor the moisture increment by Karl Fischer method after piercing the septum of a 500 mL bottle by a needle.

Measurement condition

500 mL anhydrous tetrahydrofuran (TCI product number: T2394) was compared with a competitor's 500 mL anhydrous tetrahydrofuran.

The septum moiety was pierced using an 18 gauge needle at 4 different positions every week (in total, 20 different positions pierced in 5 weeks).

After piercing, the sample solvent was taken and the water content was measured by the Karl Fischer method.

After sampling, the septum was sealed by an outer cap with 1.5 Nm torque and the sealed bottle was stored in a closed environment at *ca*. 24 °C, *ca*. 75 % RH (relative humidity).



Periodical measurement to monitor moisture increment every week (5 weeks in total) showed a remarkable difference in the water content (in ppm) between TCI and the competitor's samples. This result indicates that the PTFE sheet inside the outer cap can tightly seal the septum and the outer cap.

As a result, use of the DualSeal can maintain the quality of product in a sealed bottle for a long time.

- * Not available for smaller than the 100 mL or 100 g size bottles.
- * We are not selling the DualSeal cap itself.
- * Since the septum cap part of DualSeal contains butyl rubbers, it is not durable enough for halogenated hydrocarbon solvents such as dichloromethane. Accordingly, the minimum number of needle punctures is recommended, as it is possibly degraded by the vapor of the solvent once a hole is made.

		B4358	100mL 500mL	B4359	100mL 500mL	B4410	500mL
Volatile Solutions		CH2 CH2		CH2 CH2		CH ₃ CH ₂	
		1,3-Butad (<i>ca</i> . 15% in (liene Hexane) CAS RN: 106-99-0	1,3-Butadi (<i>ca</i> . 15% in T Ca	ene ⁻ oluene) AS RN: 106-99-0	1-Butened	(ca. 10% in Hexane) CAS RN: 106-98-9
B4411 100mL	10909 100mL 500mL	10910	100mL	10911	100mL 500mL	M2563	100mL 500mL
CH ₃ CH ₂	CH ₃ CH ₃ CH ₂	СН	CH ₃ CH ₃ CH ₂ CH ₃ CH ₃ CH ₃ CH ₂		CH ₃ CH ₃ Isoamylene		
1-Butene (<i>ca</i> . 10% in Toluene) CAS RN: 106-98-9	Isobutene (<i>ca.</i> 8% in Dichloromethane) CAS RN: 115-11-7	Isobutene (<i>ca</i> . 10% in	e Isopropyl Ether) CAS RN: 115-11-7	Isobutene (<i>ca</i> . 15% in T (Tetrahydrofuran) CAS RN: 115-11-7	rran) 5-11-7 (<i>ca.</i> 15% in Dichloromethane, <i>ca.</i> 2.5mol/L) CAS RN: 563-45-1	
M2565 100mL	P2295 100mL 500mL	P2847 100mL 500mL		M2813 100mL		T3957	100mL
CH ₃ CH ₂ Isoamylene (<i>ca.</i> 12.5% in Tetrahydrofuran, <i>ca.</i> 1.5mol/L) CAS RN: 563-45-1	CH ₃ −C≡CH Propyne (<i>ca.</i> 5% in Tetrahydrofuran, <i>ca.</i> 1mol/L) CAS RN: 74-99-7	CH ₂ Allene (ca. Tetrahydro C	CH ₂ =C=CH ₂ Allene (ca. 2% in Tetrahydrofuran, ca. 0.4 mol/L) CAS RN: 463-49-0		CH ₃ CI Methyl Chloride (ca. 5.7% in Tetrahydrofuran, ca. 1mol/L) CAS RN: 74-87-3		CF3I domethane (ca. 10% ofuran, ca. 0.5mol/L) AS RN: 2314-97-8
T3958 100mL	C2882 100mL	C2883	100mL 500mL	V0126	100mL	V0127	100mL
CF ₃ I Trifluoroiodomethane (ca. 10% in Dimethyl Sulfoxide, ca. 0.6mol/L) CAS RN: 2314-97-8	CH ₃ CH ₂ Cl Chloroethane (ca. 17% in Ethyl Ether, ca. 2.0mol/L) CAS RN: 75-00-3	CH ₃ CH ₂ Cl Chloroethane (<i>ca.</i> 15% in Tetrahydrofuran, <i>ca.</i> 2.0mol/L) CAS RN: 75-00-3		CH ₂ =CHBr Vinyl Bromide (ca. 14% in Ethyl Ether, ca. 1.0mol/L) CAS RN: 593-60-2		CH ₂ =CHBr Vinyl Bromide (ca. 12% in Tetrahydrofuran, ca. 1.0mol/L) CAS RN: 593-60-2	
M1016 500mL	M2323 100mL	M2108	100mL 500mL	E0531	100mL	E0817	100mL
CH ₃ NH ₂ Methylamine (40% in Methanol, <i>ca.</i> 9.8mol/L)	CH ₃ NH ₂ Methylamine (<i>ca.</i> 9% in Ethanol, <i>ca.</i> 2mol/L)	C Methylam Tetrahydro	H_3NH_2 hine (ca. 7% in furan, ca. 2mol/L) C AS DN 74 90 5	CH ₃ Ethylamin (30-40% in l	NH ₂ e Methanol)	CH Ethylamir (30-40% in	NH2 Ethanol)
D3948 100mL	D4198 100mL	T2704	2704 100mL		T3567 100mL 500mL		100mL
H CH ₃ CH ₃ CH ₃ Dimethylamine (<i>ca.</i> 10% in Tetrahydrofuran, <i>ca.</i> 2mol/L)	H CH₃ ^{× N} CH₃ Dimethylamine (ca. 11% in Alcohol, ca. 2mol/L)	CH Trimethyl Tetrahydro	CH_3 $^{N}CH_3$ amine (ca. 13% in furan, ca. 2mol/L)	CH ₃ Trimethyla Acetonitrile	CH_3 N CH_3 mine (ca. 13% in , ca. 2mol/L)	CH Trimethyl Toluene, ca	CH ₃ I I ₃ CH ₃ amine (<i>ca.</i> 8% in <i>i</i> . 1mo//L)
CAS RN: 124-40-3	CAS RN: 124-40-3	A2236	100ml 500ml	A2237	100ml	H1060	CAS RIV: 75-50-3
CH ₃ CH ₃ CH ₃ CH ₃ CH ₃	NH ₃		NH ₃		NH ₃		HCI
Irimethylamine (<i>ca</i> . 25% in Methanol, <i>ca</i> . 3.2mol/L) CAS RN: 75-50-3	Ammonia (<i>ca.</i> 4% in Methanol, <i>ca.</i> 2.0mol/L) CAS RN: 7664-41-7	Ammonia (ca. 4% in Ethanol, ca. 2.0mol/L) CAS RN: 7664-41-7		Ammonia Alcohol, <i>ca</i> . CA	(<i>ca</i> . 4% in Isopropyl 2.0mol/L) S RN: 7664-41-7	Hydroger (ca. 1mol/L	n Chloride . in Ethyl Acetate) AS RN: 7647-01-0
H1061 100mL	H1062 500mL	H1277 500mL		X0041 100mL 500mL		H0959 100mL 500mL	
HCI Hydrogen Chloride (ca. 1mol/L in Ethyl Ether) CAS RN: 7647-01-0	HCI Hydrogen Chloride (ca. 4mol/L in 1,4-Dioxane) CAS RN: 7647-01-0	Hydroger (ca. 16% in Ether, ca. 4 C,	HCI n Chloride Cyclopentyl Methyl mol/L) AS RN: 7647-01-0	Hydrogen Methanol [for Esterific CA	Chloride - Reagent (5-10%) astion] S RN: 7647-01-0	Hydroger Ethanol R [for Esterifi CA	n Bromide - eagent (10-20%) ication] S RN: 10035-10-6

U0147	100mL	U0148 100m	T2346 10	0mL 500mL		
SO ₂ Sulfur Dioxide (ca. 8 Tetrahydrofuran, ca. 1 CAS RN: 744	% in .2 mol/L) 16-09-5	SO ₂ Sulfur Dioxide (ca. 2.5% in Dichloromethane, ca. 0.5 mol/ CAS RN: 7446-09-	Tetrahydrofuran I in Tetrahydrofuran, (stabilized with Sod CAS RN:	• BH ₃ Borane (8.5% <i>ca</i> .0.9mol/L) lium Borohydride) • 14044-65-6		
Metallie	c Sa	It Solutions	L0186 LiBl Lithium Boroh (ca. 4mol/L in Te CAS RN	100mL H ₄ hydride htrahydrofuran) I: 16949-15-8	LiCl LiCl Lithium Chloride (2.3% in Tetrahydrofuran, ca. 0.5mol/L) CAS RN: 7447-41-8	TiCl ₄ Ticli4 Titanium(IV) Chloride (14% in Dichloromethane, ca. 1.0mol/L) CAS RN: 7550-45-0
SnCl ₄ Tin(IV) Chloride (ca. in Dichloromethane) CAS RN: 764	1.0mol/L 16-78-8	ZnCl ₂ Zinc Chloride (<i>ca.</i> 7% in Tetrahydrofuran, <i>ca.</i> 0.5mol// CAS RN: 7646-85-	ZINC ZINC ZINC Chloride 2-Methyltetrahy ca. 2mol/L) CAS RI	100mL 212 (ca. 25% in rdrofuran, N: 7646-85-7	H1197 500ml	D3479 500ml
Dehyd	rate	d Solvents	CH2 Dichlorometh Anhydrous (st. 2-Methyl-2-bute CAS	Cl ₂ ane abilized with ene) i RN: 75-09-2	CH ₃ CH ₃ Hexane Anhydrous CAS RN: 110-54-3	CH ₃ OCH ₃ Diethyl Ether Anhydrous (stabilized with BHT) CAS RN: 60-29-7
Tetrahydrofuran Ani (stabilized with BHT) CAS RN: 10	500mL hydrous)9-99-9					
Orga R	ano leag	metallic Jents	M0362 CH ₃ M Methylmagnes (12% in Tetrahydr <i>ca</i> . 1mol/L) CAS	250g IgBr ium Bromide rofuran, 5 RN: 75-16-1	E0134 250g CH ₃ MgBr Ethylmagnesium Bromide (39% in Ethyl Ether, <i>ca.</i> 3mol/L) CAS RN: 925-90-6	B1933 250g CH ₂ MgCl Benzylmagnesium Chloride (ca. 16% in Tetrahydrofuran, ca. 1mol/L) CAS RN: 6921-34-2
CH ₃ n-Octylmagnesium E (ca. 22% in Tetrahydrofu ca. 1mol/L) CAS RN: 1704 T1533	250g MgBr Bromide Jran, 19-49-9 100mL	S0467 100g 500g NaAl(OCH2CH2OCH3)2H2 Sodium Bis(2-methoxyethoxy aluminum Dihydride (70% in Toluene, ca. 3.6mol/L) CAS RN: 22722-98- D3214 100mL 500m	B1087 CH ₃ -S CH ₃ -S C Bromotrimeth CAS R	250mL CH ₃ Si-Br CH ₃ nylsilane N: 2857-97-8 100mL	T1451 100mL CH3 I CH3-Si-CH2MgCl I CH3 CH3 Trimethylsilylmethylmagnesium Chloride (20% in Ethyl Ether, ca. Imol/L) CAS RN: 13170-43-9 H0915 100mL 500mL	T0585 100g Image: Normal System Image: Normal System<
(CH ₃) ₂ CH—Si—H (CH ₃) ₂ CH—CH(C CH(C Triisopropylsilane CAS RN: 648	H H ₃) ₂ 35-79-6	Zn(CH ₂ CH ₃) ₂ Diethylzinc (ca. 15% in Hexane, ca. 1mol/L) CAS RN: 557-20-1	Zn(CH ₂ Diethylzinc (cc Toluene, ca. 1m CAS I	CH ₃) ₂ 7. 15% in ol/L) RN: 557-20-0	CH ₃ -Si-N-Si-CH ₃ CH ₃ -Li CH ₃ Li CH ₃ LiHMDS (<i>ca.</i> 26% in Tetrahydrofuran, <i>ca.</i> 1.3mol/L) CAS RN: 4039-32-1	CH ₃ —Si—N—Si—CH ₃ CH ₃ Na CH ₃ NaHMDS (contains 2-Methyl-2-butene) (38% in Tetrahydrofuran, <i>ca.</i> 1.9mol/L) CAS RN: 1070-89-9

DualSeal — Airtight-Double-Cap Bottled Reagents



Boron Trifluoride - Butyl Ether Complex (BF₃ *ca.* 30%) CAS RN: 593-04-4

- * Not available for smaller than the 100 mL or 100 g size bottles.
- * We are not selling the DualSeal cap itself.
- * Since the septum cap part of DualSeal contains butyl rubbers, it is not durable enough for halogenated hydrocarbon solvents such as dichloromethane. Accordingly, the minimum number of needle punctures is recommended, as it is possibly degraded by the vapor of the solvent once a hole is made.

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